

Coating of nitinol surface by titanium dioxide films grown by atomic layer deposition



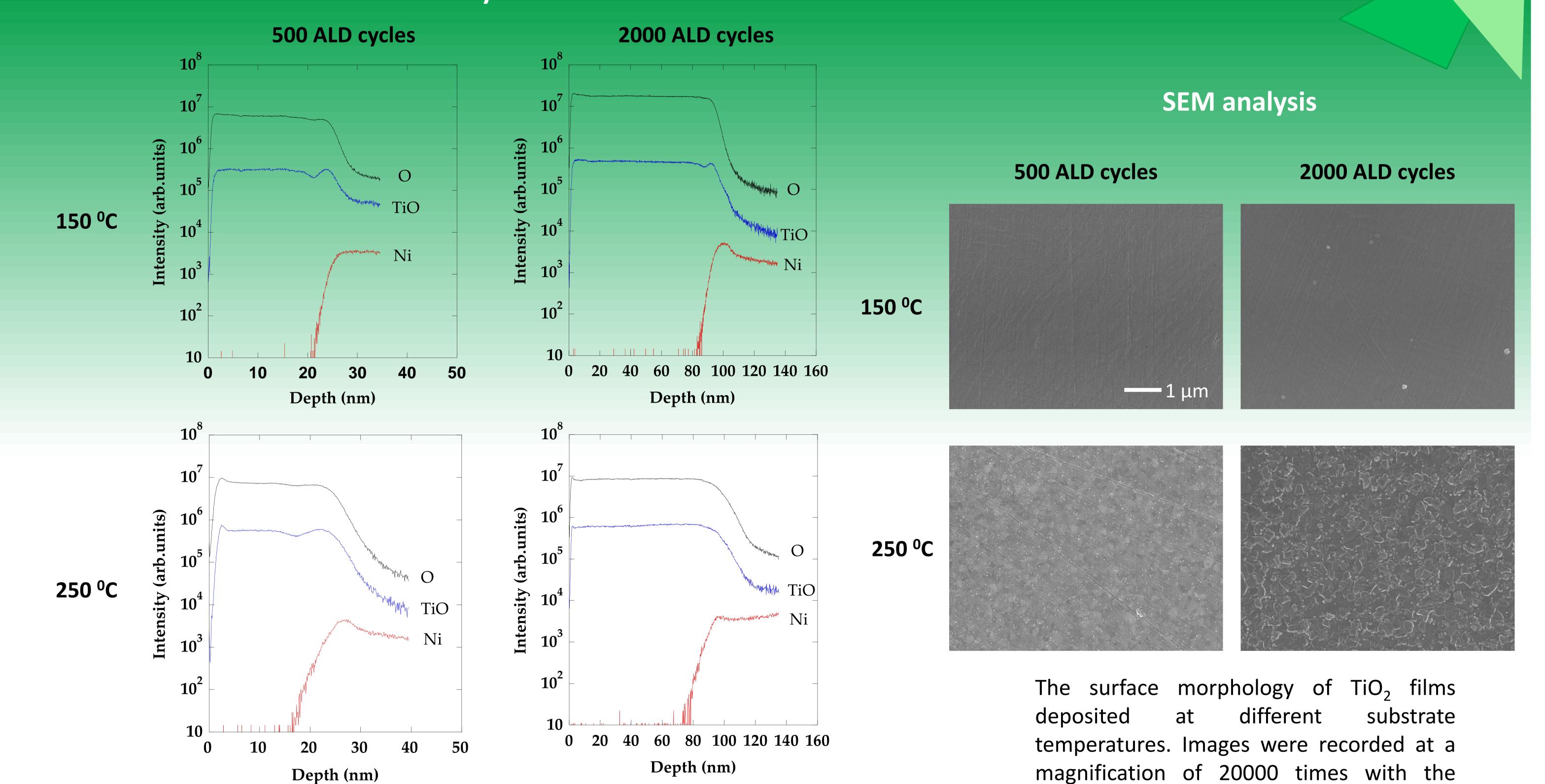
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Introduction

Nitinol (NiTi), a nearly equiatomic alloy of Ni and Ti atoms, with its unique features such as shape memory, superelasticity and kink resistance, has found many applications in biomedical device production. However, the release of Ni from nitinol is of considerable concern, as Ni atoms are known to be allergenic and toxic. Therefore, modification and coating of NiTi is highly desirable in order to improve the biocompatibility and corrosion resistance of material and prevent the release of Ni into human body.

Titanium dioxide (TiO₂) is a perfect candidate for the biocompatible coatings on biomedical materials, due to its biological and chemical inertness, nontoxicity and long-term stability against corrosion. In this work we have used atomic layer deposition (ALD) technique for the synthesis of thin TiO₂ films on NiTi substrates. Films were deposited in a wide temperature range (150 – 250 °C), using TiCl₄ and water as ALD precursors. SIMS spectromety was employed for the investigation of Ni diffusion into TiO₂, while the surface morphology and the chemical composition of deposited films were analysed with SEM microscopy and XPS spectroscopy.

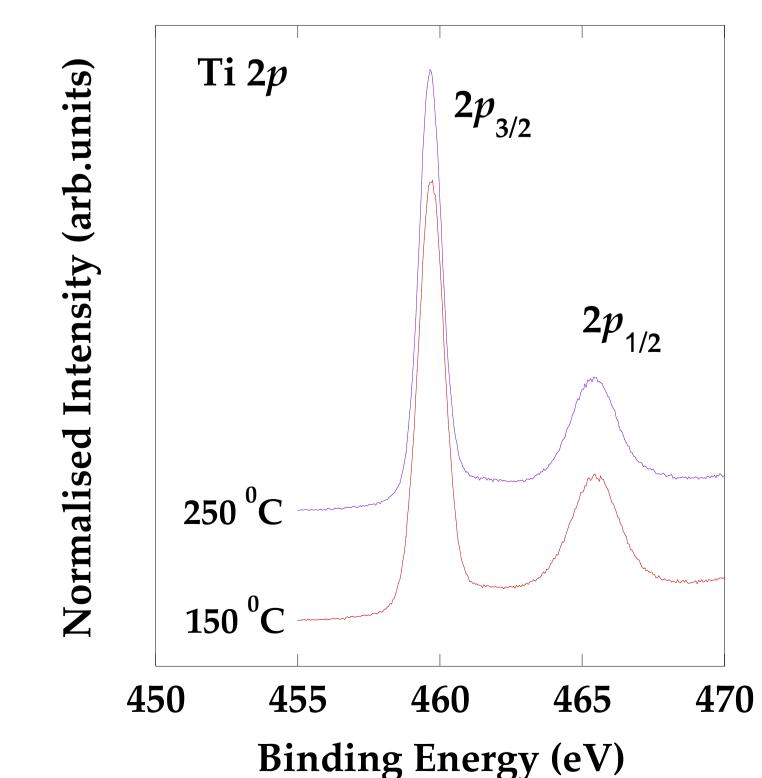
SIMS analysis



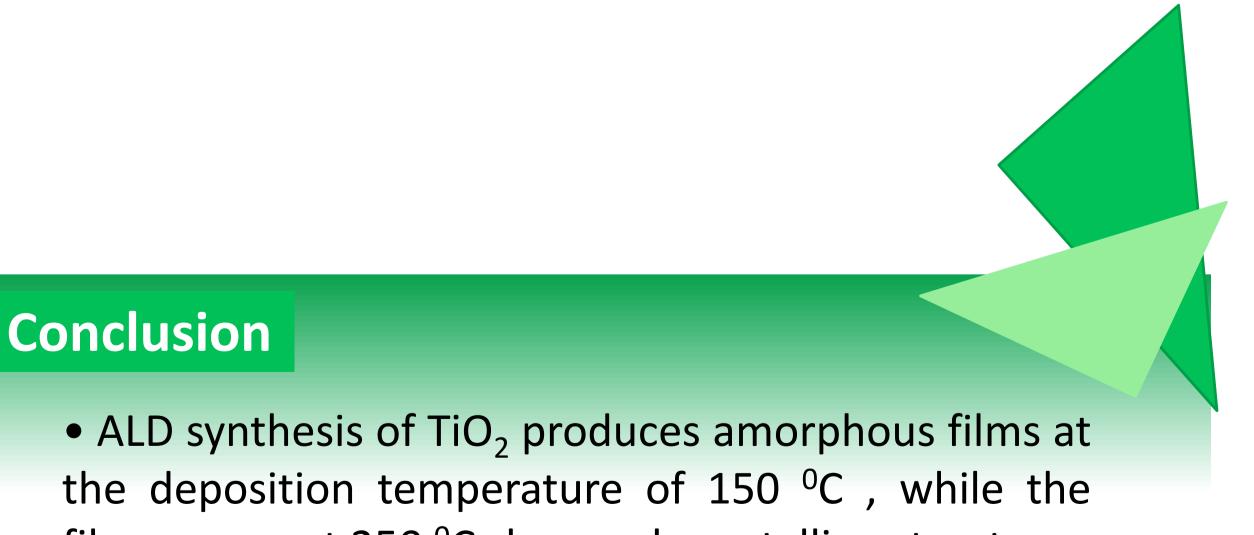
In-depth SIMS profiles of TiO₂ films recorded by measuring the negative ions of O, Ni and TiO (with 5 keV Cs⁺ primary ion beam). Diffusion of Ni into the TiO₂ film is more pronounced in the films deposited at 250 °C (polycrystalline films), compared to films grown at 150 °C (amorphous films).

XPS analysis

XPS spectra measured around Ti 2*p* core levels, for samples deposited at 150 °C and 250 °C, show a structure characteristic for a pure TiO_2 .



magnification of 20000 times with the gentle electron beam of 0.7 keV. Films deposited at 150 ^oC show amorphous structure, in contrast to polycrystalline films grown at 250 °C.



2000 ALD cycles



 SIMS measurements show that an amorphous phase of deposited films act as a better diffusion barrier for the release of Ni in TiO₂, compared to polycrystalline films.

